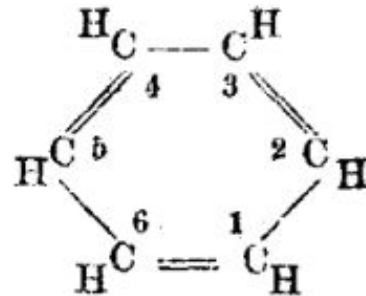
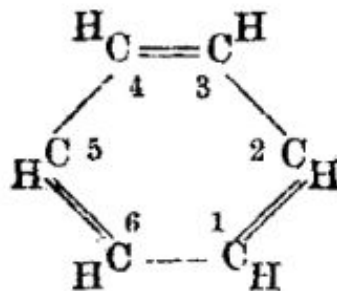


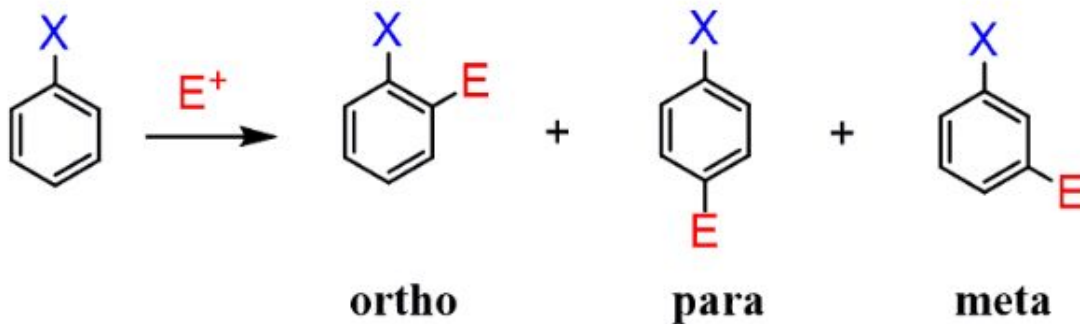
# Benzene



Molecular formula:  $C_6H_6$

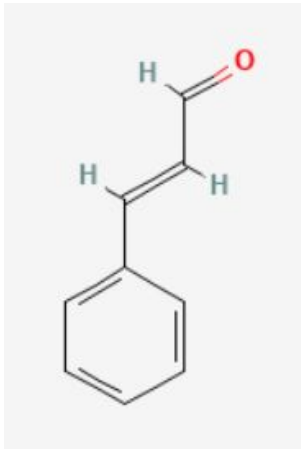
Faraday, Couper, Loschmidt, Kekulé

Aromatic, aryl, phenyl

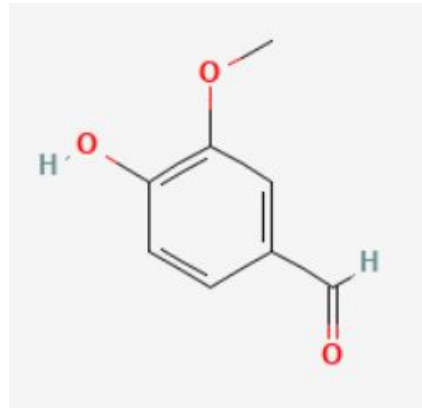


# Why are they called aromatics?

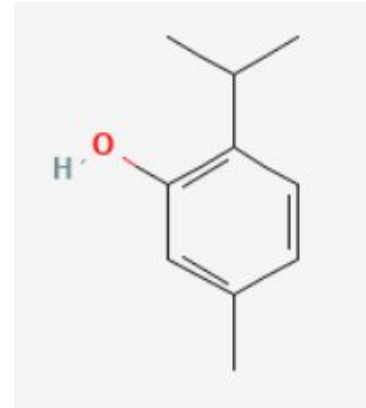
- Various aromas associated with them during early discovery



Cinnamaldehyde



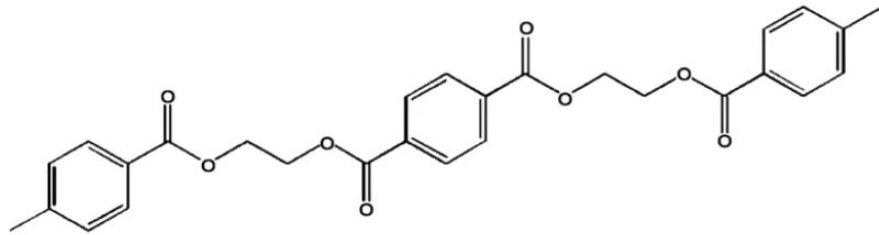
Vanillin



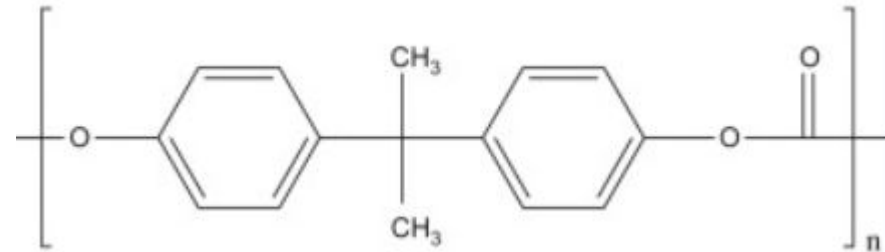
Thymol

# Conjugation

- Delocalization of electrons, freely move throughout a system
- Principle of electronics
- Organics can provide conducting and semiconducting materials
- Used in plastics, solar cells, thin films



Polyethylene terephthalate (PET)



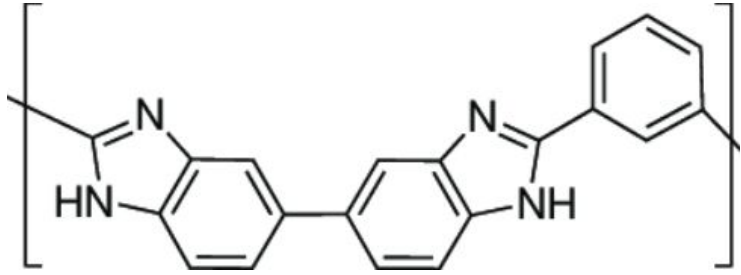
Polycarbonate (PC)

# Plastics

- Many commodity plastics are aromatic polymers



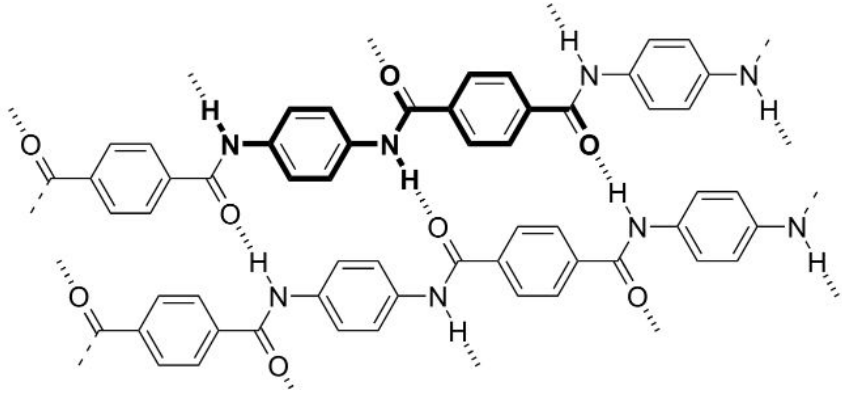
# Materials



polybenzimidazole



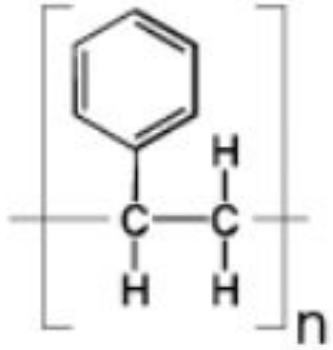
# Materials



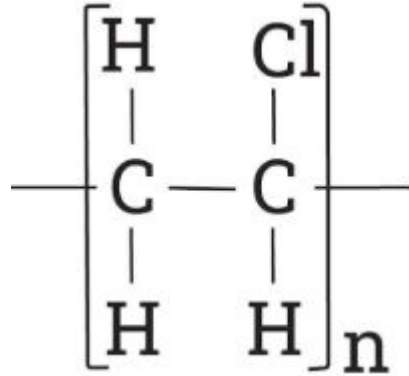
Kevlar (para-aramid)



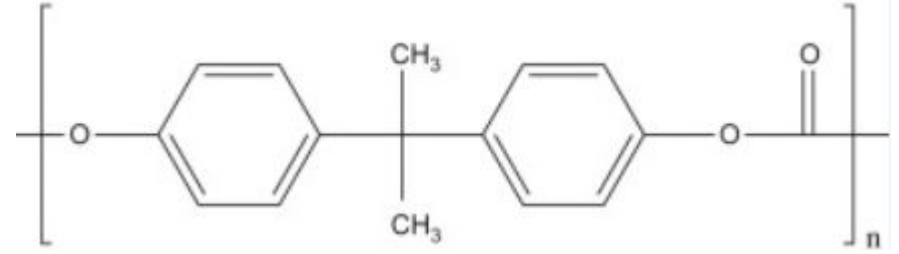
# Recyclability of Polymers



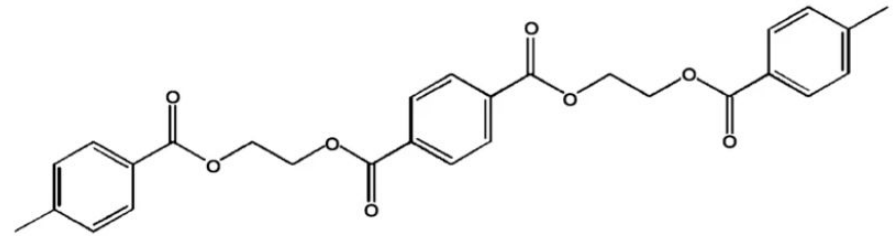
Polystyrene (PS)



Polyvinyl chloride (PVC)



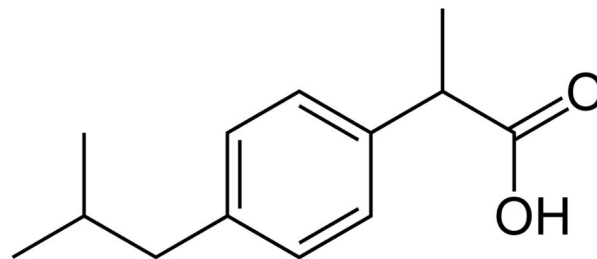
Polycarbonate (PC)



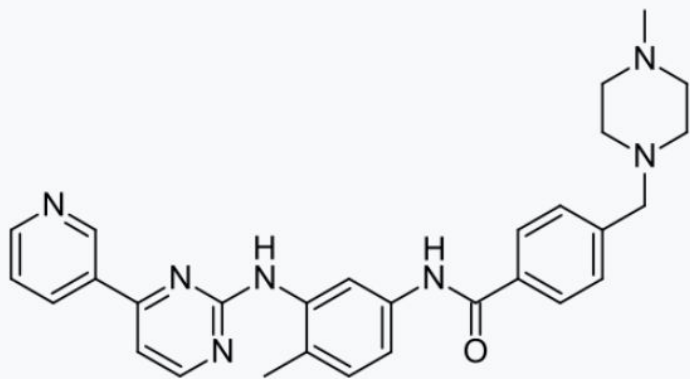
Polyethylene terephthalate (PET)

# Pharmaceuticals

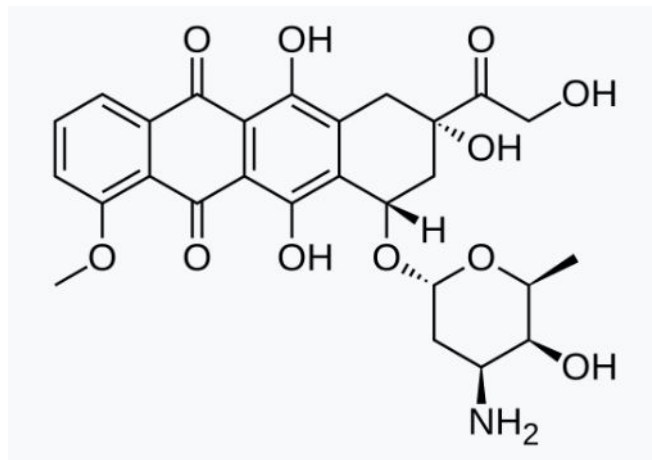
- Over 75% are aromatic!



ibuprofen



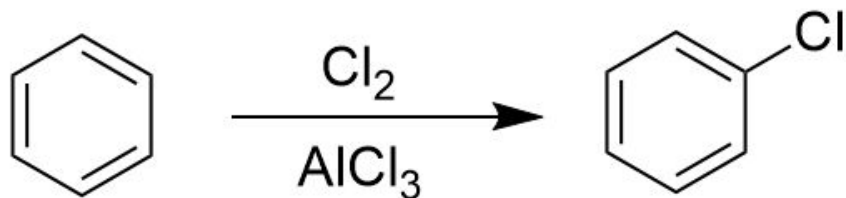
imatinib



doxorubicin

# Substitution Reactions

halogenation



nitration

