

Name \_\_\_\_\_

Date \_\_\_\_\_ Period \_\_\_\_\_

## A Brief History of the Atom

Directions: Please fill out the following guided notes as we watch a short video describing the history of the atom.

The first recorded development in Atomic History can be found over \_\_\_\_\_ years ago in Ancient Greece. The philosophers \_\_\_\_\_ and \_\_\_\_\_ are considered to be the two co-originators of the idea that all matter consists of **indivisible** particles called \_\_\_\_\_. This is Greek term “indivisible” and is where we get the word \_\_\_\_\_.

Democritus’ ideas surrounding the atom were rejected and ignored by the \_\_\_\_\_, government, and society in general until the 1600s. That’s over 2000 years!

In 17<sup>th</sup> century Europe, the \_\_\_\_\_ Method was developed which **required scientists to provide evidence to support their claims and ideas**. In \_\_\_\_\_, an English chemist named John Dalton published his ideas about Atoms with support from his research. He proposed the following:

- All matter is comprised of \_\_\_\_\_.
- These are held together by forces of attraction.
- All atoms are \_\_\_\_\_.
- Atoms cannot be \_\_\_\_\_ nor \_\_\_\_\_.
- Atoms in the same \_\_\_\_\_ are identical and have the same \_\_\_\_\_.
- Atoms of different elements combine to form \_\_\_\_\_.

In the next 100 years, it was found that Dalton’s proposals were not entirely correct. In particular, atoms are NOT \_\_\_\_\_.

In the early 1900’s, physicist JJ Thomson concluded that there are two parts to atoms:

1. Heavy part with a \_\_\_\_\_ charge
2. Negatively charged particles called \_\_\_\_\_

He described the structure of atoms by comparing it to a \_\_\_\_\_.

In 1911, one of Thomson’s students named Ernest Rutherford extended this research by firing positively-charged alpha particles into a thin sheet of \_\_\_\_\_ foil, and developed the **Nuclear Model** of the Atom.

This model describes atoms as having a **positively charged core** called the \_\_\_\_\_, and that **electrons** moved in random orbits around it.

In 1912, Neils Bohr concluded that electrons revolve around a center nucleus, but at certain distances away. These electron orbits are called \_\_\_\_\_.

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## The History of the Atom - Main Points

Democritus

400 BC - Ancient Greece



John Dalton

1803-1805



JJ Thomson

1897



Ernst Rutherford

1909-1911



Neils Bohr

1912-1913

