Chemistry 2, HW 22

Look at the slides 2-5 in the class presentation, look at organic functional group interconversions, find how haloalkane can be converted to alkenes (the same reaction will work for haloalkane to alkyne conversion), this will be helpful for questions 1 and 3. For question number 2, you have to remember reaction characteristic for benzene.

Also keep in mind, that though alkanes are not reactive, but they will react with halogens in the presence of sunlight or ultraviolet light.

$$CH_4 + Cl_2 \rightarrow CH_3Cl + HCl$$

Questions:

Try to figure out the following conversions

1.
$$C_2H_4 \rightarrow C_2H_4Br_2 \rightarrow C_2H_2$$

2. $C_6H_6 \rightarrow C_6H_5Br \rightarrow C_6H_5ONa$
3. P_{\bullet}

Where x – intermediate unknown compound

4.
$$C_2H_6 \rightarrow X \rightarrow C_2H_5OH$$