

- $2(\text{H}_2\text{O})_2$ In this formula subscript tell you = Number of atoms
- $2(\text{H}_2\text{O})_2$ the coefficient tells you = Number of molecules

Chemical formula/Compound	Element 1	Element 2	Element 3	Total
CaCO₃	Ca=	C=	O=	
2CaSO₄				
3Ca(NO₃)₂				
2Mg(OH)₂				
2K₂Cr₂O₇				
2CH₄				
2Al₂(SO)₃				
Na₃PO₄				
2Na₂CO₃				