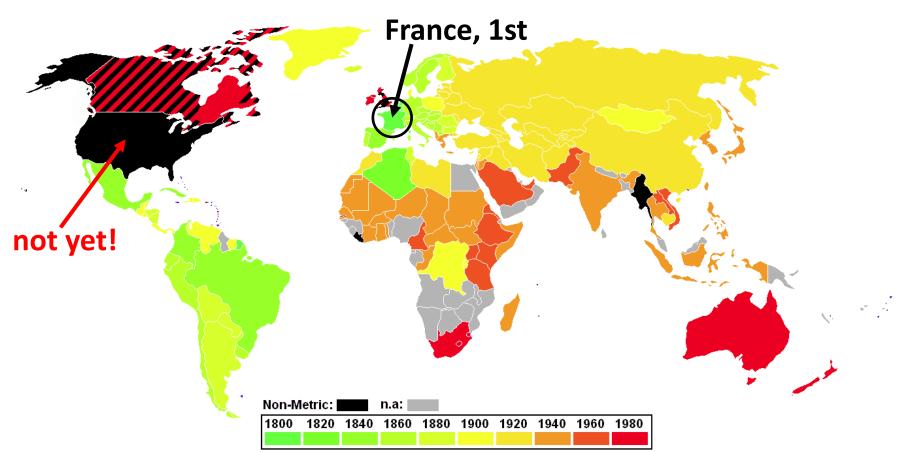
Metrication of the World



Currently USA is the only country (and perhaps also Myanmar and Liberia) that has not fully adopted the Metric System as its official system of measurement...as a result, Metric System is used in Science, but not Manufacturing!

Loss of NASA orbiter

NASA's Mars Climate Orbiter lost on September 23, 1999. Cost: \$125 million.

For a <u>key spacecraft operation</u>, Lockheed Martin engineering team used Imperial units of measurement while the NASA's team used more conventional Metric system...

The spacecraft insertion trajectory came <u>too close to the planet</u>; the Orbiter <u>disintegrated</u> upon entering the upper Martian atmosphere.





Conversion of Units

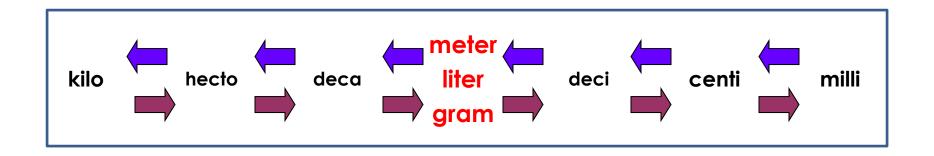
 For <u>the same quantity measured</u>, we can convert units using an <u>equivalence statement</u> which shows the relationship between the units (this relationship is called a *conversion factor*).

Imperial-Metric equivalence statements:

- Units of LengthUnits of WeightUnits of Capacity> 1 in = 2.54 cm> 1 oz = 28.35 g> 1.06 qt = 1 L> 3.28 ft = 1 m> 1 lb = 454 g> 1 gal = 3.79 L> 1 mi = 1.61 km> 2.2 lb = 1 kg
- Units that measure *physical quantities* (like the examples above) always have a common zero.

Within the Metric System itself,

by design, conversion factors are always a power of 10.



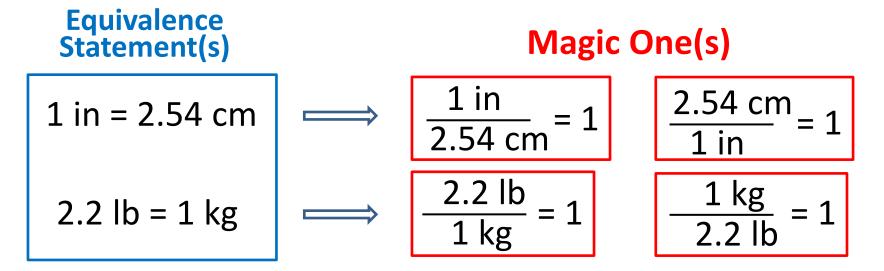
- If you convert into larger units (towards the left in the diagram), move the decimal point to the left.
 4000000 millimeters = 4.00000 kilometers = 4 kilometers
- If you convert into smaller units (toward the right in the diagram), move the decimal point to the right.

7 kilograms = 7.000000 kilograms = 7000000 milligrams

Dimensional Analysis



- <u>Dimensional Analysis</u> (also called Factor-Label Method or the Unit Factor Method) is a problemsolving method that uses the fact that any number or expression can be <u>multiplied by 1</u> (Magic One) without changing its value.
- To help with conversion of units, Magic One is built using the equivalence statement:



Example: Convert 130 lbs to kg

Step 1. Write the original measurement as a unit fraction: 130 lbs / 1

Step 2. Using the equivalence statement, build a magic one (building rule - the numerator unit is the unit you want, the denominator unit is the original unit you want to eliminate):

2.2 lb = 1 kg
$$\longrightarrow \frac{1 \text{ kg}}{2.2 \text{ lb}} = 1$$

Step 3: multiply your unit fraction by your magic one and write your answer in the new units:

$$\frac{130 \text{ lbs}}{1} \cdot \frac{1 \text{ kg}}{2.2 \text{ lbs}} = \frac{130 \text{ kg}}{2.2} = 59.1 \text{ kg}$$

Example: The fuel tank of a plane can hold 876 liters of gas. How many gallons would it be?



Equivalency: 1 gallon = 3.8 liters

$$\frac{876 \text{ L}}{1} \cdot \frac{1 \text{ gal}}{3.8 \text{ L}} = \frac{876 \text{ gal}}{3.8} = 230.5 \text{ gal}$$

Gimli Glider

July 23, 1983: Air Canada Flight 143 (Boeing 767-233 jet), ran out of fuel at an altitude of 41,000 feet (12 km), about halfway through its flight from Montreal to Edmonton.

The crew were able to glide the aircraft safely to an emergency landing at Gimli Industrial Park Airport. <u>None of the 61</u> <u>passengers were seriously hurt</u>.

Investigation: fuel loading was miscalculated due to a misunderstanding of the recently adopted metric system which replaced the imperial system.







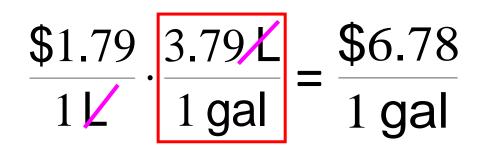
Exercise: As a practical joke, for the Candid Camera show, a gas station listed their price as \$1.79/L. People gassing up thought they were getting a great deal, but then were outraged when their total owed came up. WHY?

What should we do?



Let's carefully examine: "Listed their price as \$1.79/L"

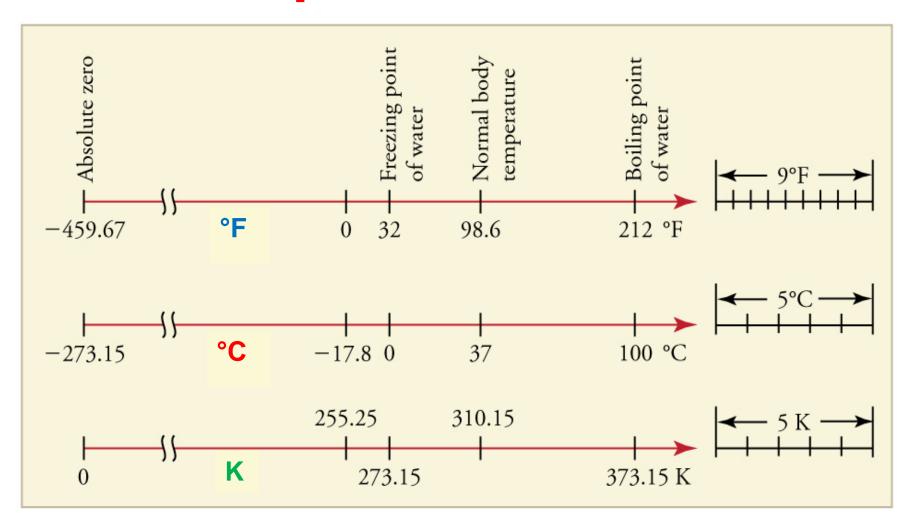
Equivalency: 1 gal = 3.79 L



"The deal" was actually **\$6.78/gal**!



Temperature Scales



Note: according to the latest research, <u>normal human</u> <u>body temperature</u> is 36.8 °C ±0.7 °C, or 98.2 °F ±1.3 °F.

Conversion of Temperature

When converting temperature between different scales, we need to pay attention to the fact that they all have <u>different "0" points</u>, therefore not only a *multiplication factor* is needed but also a *shift*.

<u>Kelvin</u>

K = °C+273.15

Fahrenheit

$${}^{\mathsf{P}}\mathsf{F} = {}^{\mathsf{o}}\mathsf{C} \cdot 1.8 + 32 = {}^{\mathsf{o}}\mathsf{C} \cdot \frac{9}{5} + 32$$

Celsius (Centigrade) °C = (°F-32)/1.8 = (°F-32) $\cdot \frac{5}{9}$