

# Length scales in Nature

1 mm



Grain of sugar, small insects, etc

1 km



Brooklyn bridge

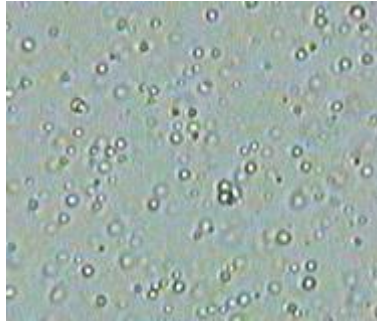
$10^{-3}$  m

1 m

$10^3$  m

1 micron ( $1\mu\text{m}$ )

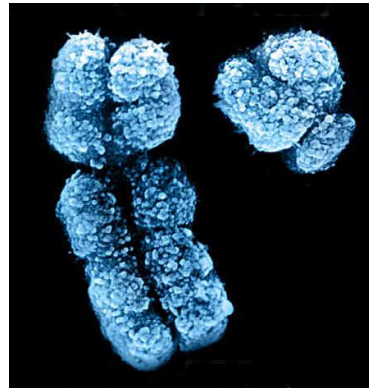
Particles in smoke, milk, etc  
( $1\text{--}20\mu\text{m}$ )



Bacteria  
( $1\text{--}10\mu\text{m}$ )



Human Chromosome  
( $2\text{--}10\mu\text{m}$ )



1000 km



$10^{-6}$

$10^{-3}$

1 m

$10^3$

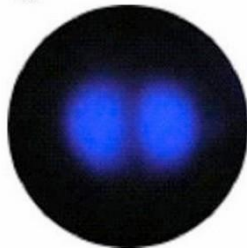
$10^6$

**1 nanometer = 10 Angstrom**  
**(1 nm = 10 Å)**

a

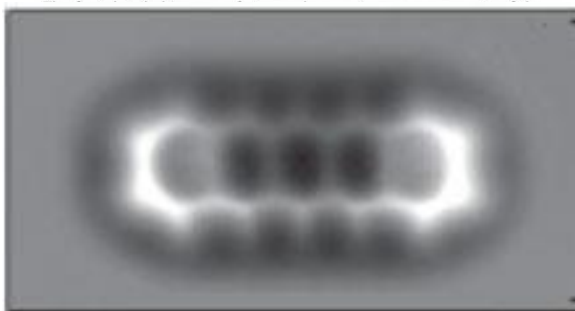


b

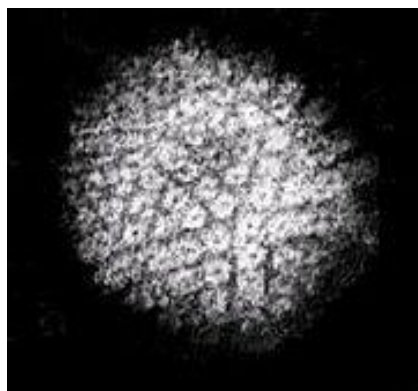


**Atom (1 Å)**

**Molecule  
(1nm)**



**Virus (>10 nm)**



**1,000,000 km**  
**(3 light seconds)**



$10^{-9}$

$10^{-6}$

$10^{-3}$

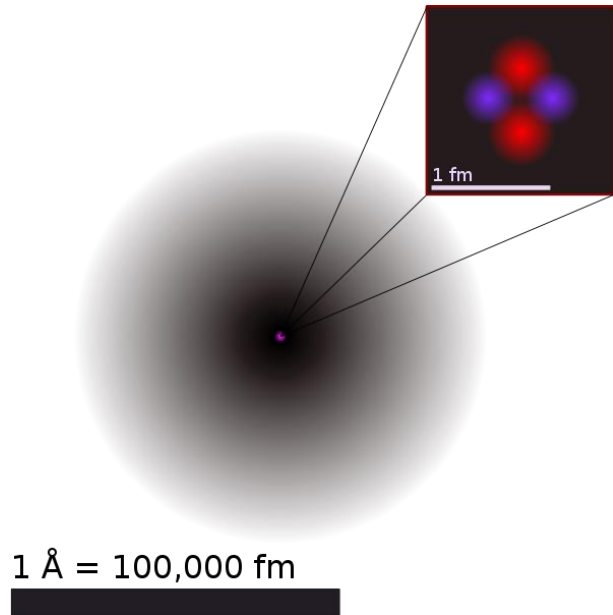
1 m

$10^3$

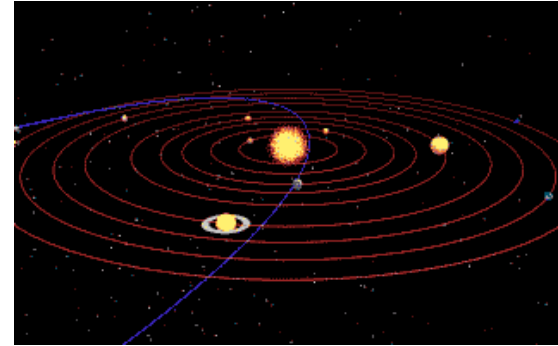
$10^6$

$10^9$

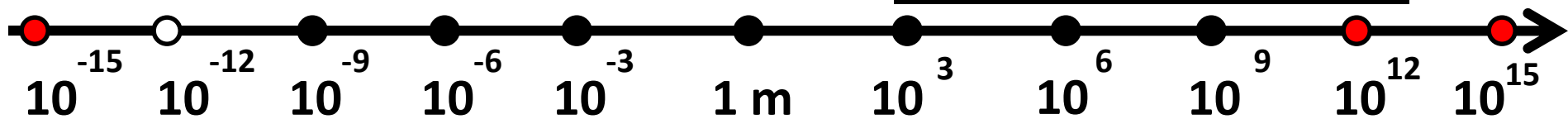
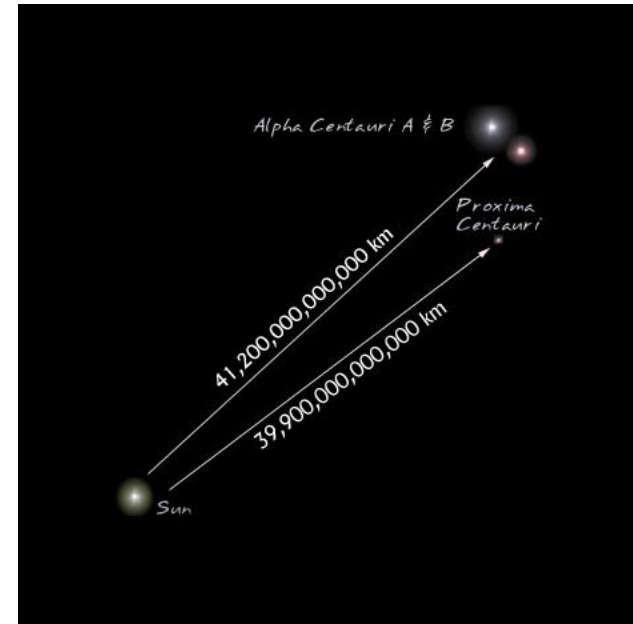
# Proton, neutron, atomic nucleus



$10^{12} \text{ m} = 1 \text{ billion km} \approx 1 \text{ light hour}$



$10^{16} \text{ m} \approx 1 \text{ light year}$

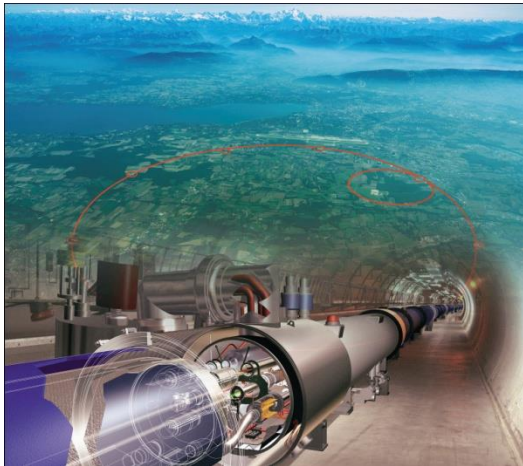


# Modern Physics

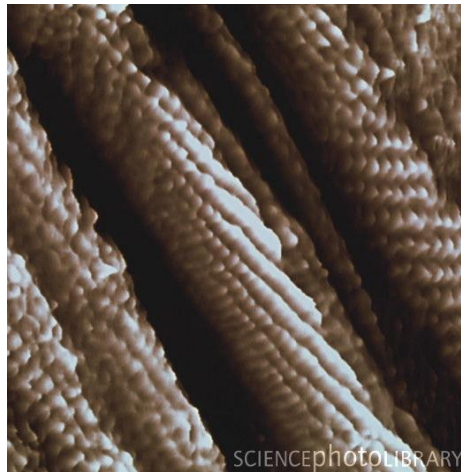
High Energy Physics

Condensed Matter Physics

Astrophysics & Cosmology



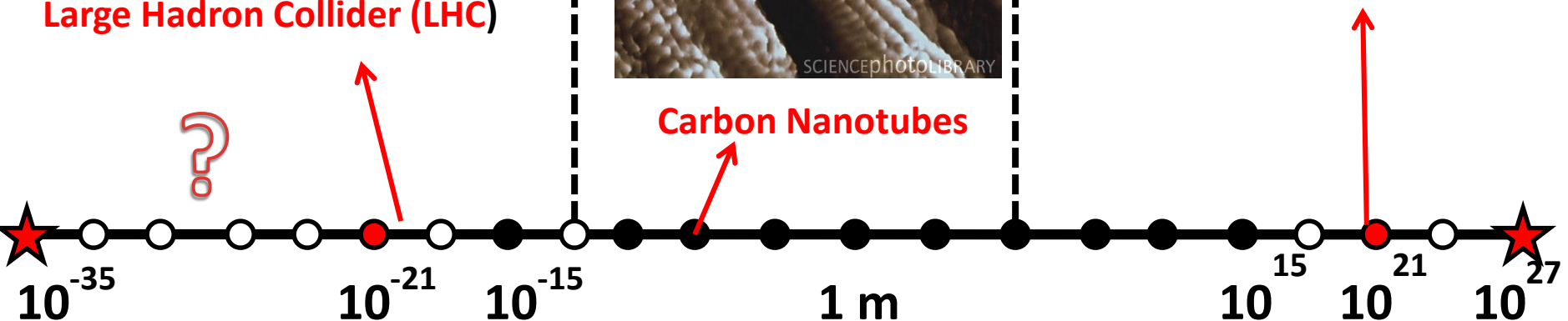
Large Hadron Collider (LHC)



Carbon Nanotubes



Our Galaxy (Milky Way)





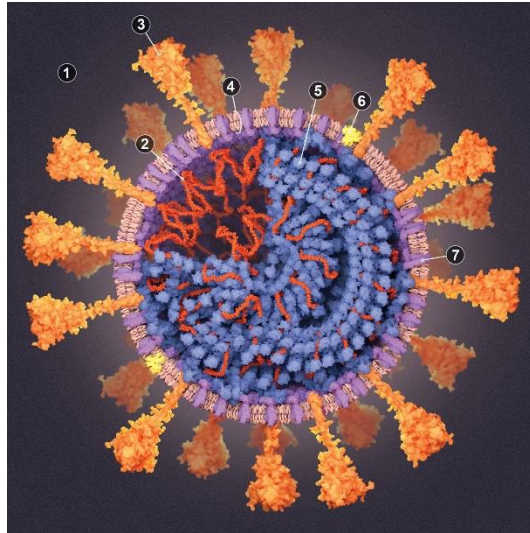
# The difference between droplet and airborne transmission

## Droplet transmission

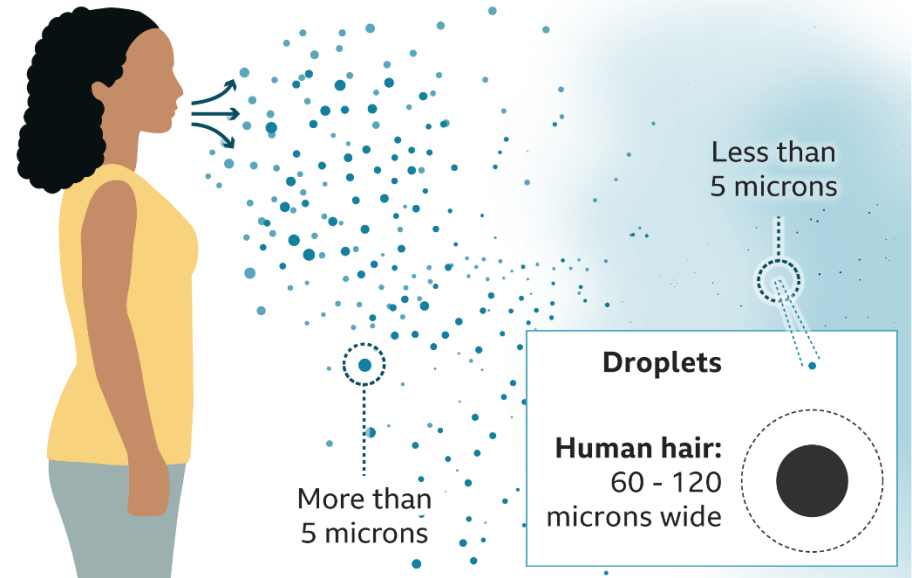
Coughs and sneezes can spread droplets of saliva and mucus

## Airborne transmission

Tiny particles, possibly produced by talking, are suspended in the air for longer and travel further



about 100 nm = 0.1  $\mu$ m



Source: WHO

BBC

pore size: < 0.3  $\mu$ m

1..10  $\mu$ m

about 10  $\mu$ m

about 100  $\mu$ m (0.1 mm)



**N95**



**Surgical masks**



**Cotton masks**



**Cloth coverings**

# Homework 1

## Problem 1.

Water molecule can be approximated as a sphere of radius  $2 \text{ \AA}$  ( $1 \text{ \AA} = 10^{-10} \text{ m}$ , is called Angstrom). Estimate how many molecules is there in  $1 \text{ cm}^3$  of water.

## Problem 2.

Estimate the number of cells in your body, by approximating a single cells as a cube sized  $10 \times 10 \times 10$  micron. **Hint:** if you know your mass, you know your volume.