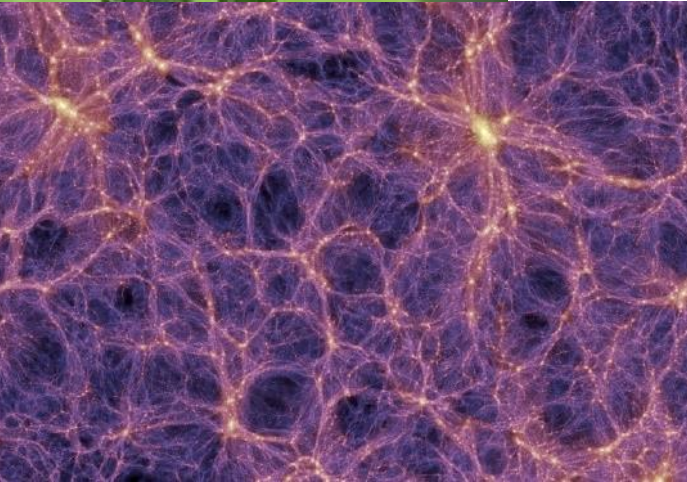




OH DEAR, WHAT CAN THE
MATTER
BE?



What is Matter?

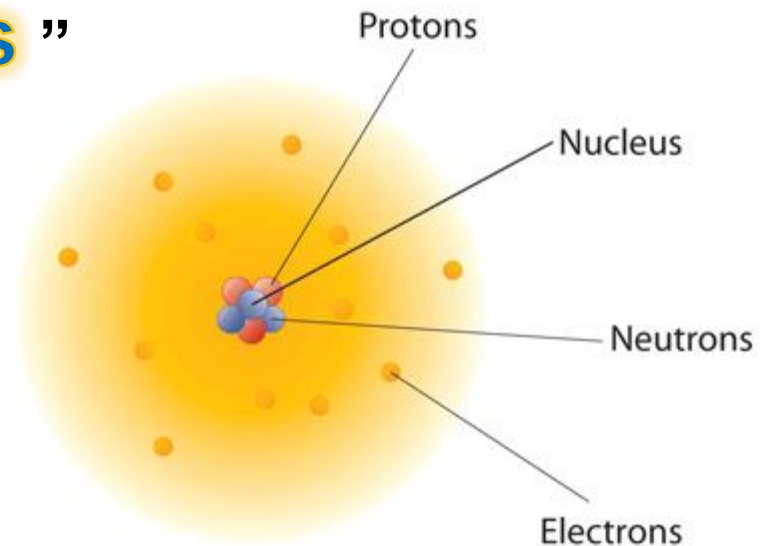
1. Common “classical” definition (known as *mechanical, abstract mathematical*), **René Descartes, Isaac Newton** - 17th century:

“**Matter is anything that has mass and takes up space**”

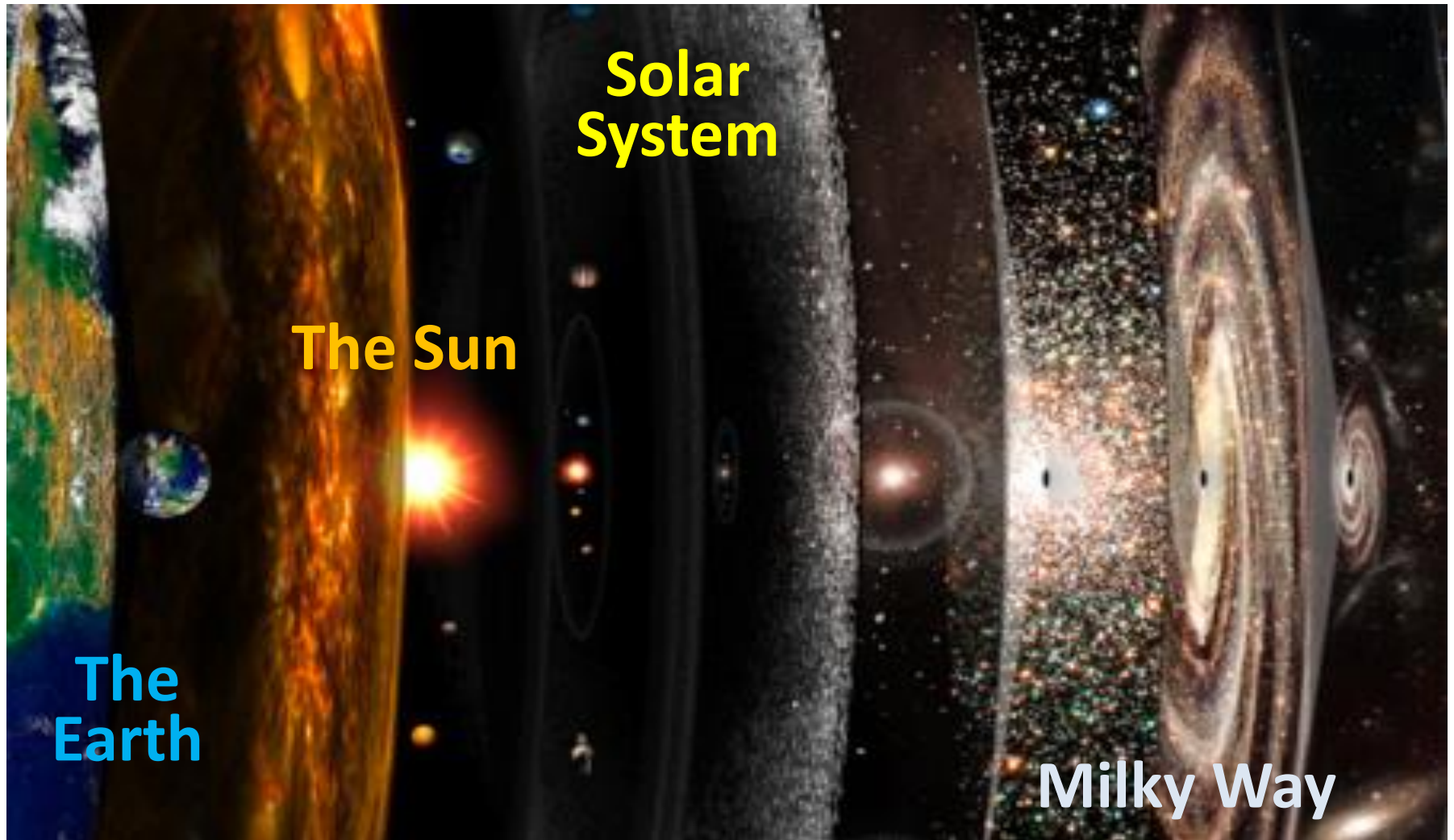
2. Late 19th century definition (based on physical and chemical structure):

“**Matter is made up of atoms**”

This *atomic, or ordinary,* matter is in turn made up of interacting *subatomic particles* — usually a nucleus of *protons* and *neutrons*, and a cloud of orbiting *electrons*.



Some Bigger (>1 million m) Things



The Earth

10^7 m

The Sun

10^9 m

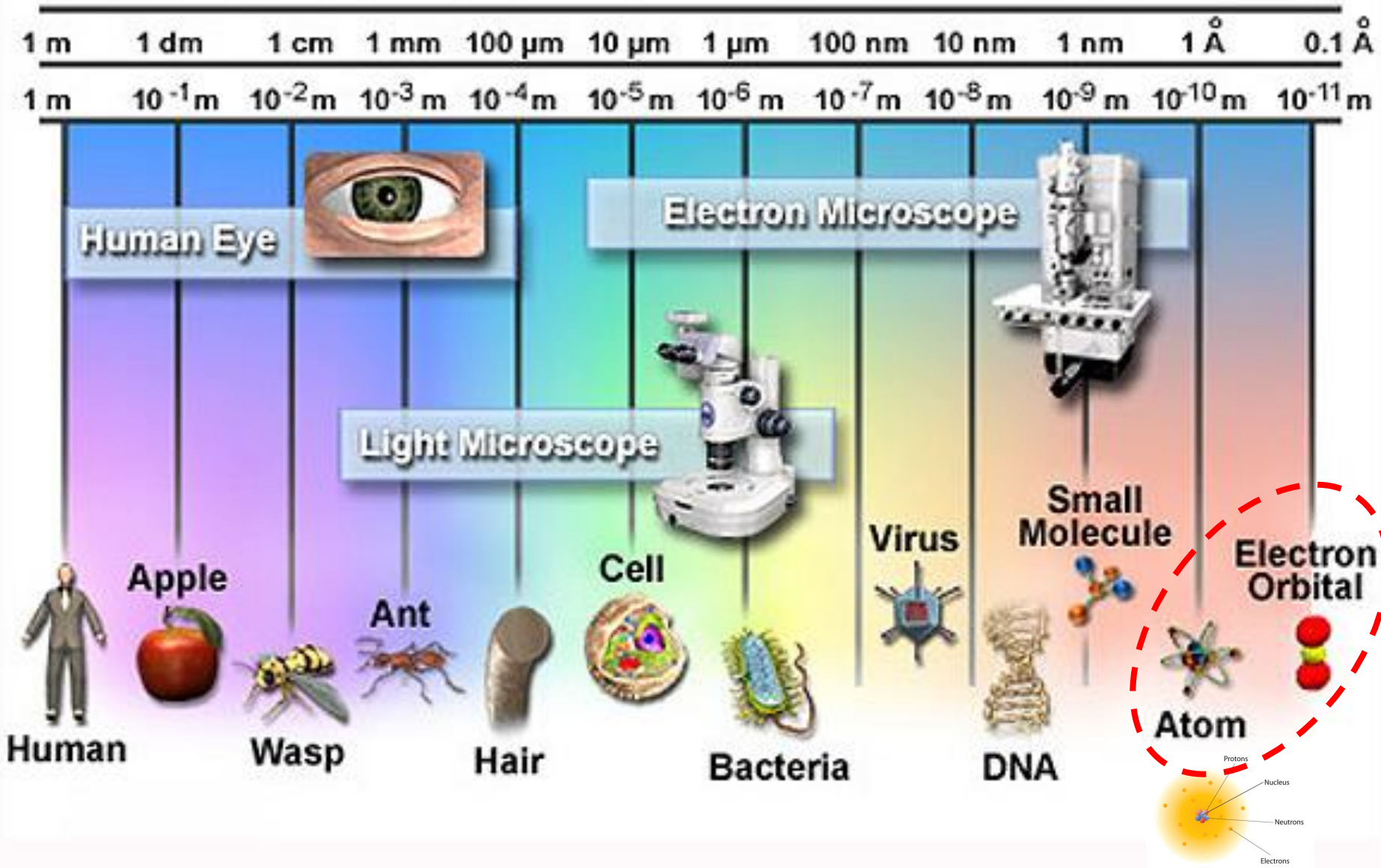
Solar System

10^{13} m

Milky Way

10^{21} m

Some Smaller (<1 m) Things



Voyage into the World of Atoms



THERE ARE
MORE ATOMS IN
A SINGLE
GRAIN OF SAND
THAN GRAINS
OF SAND ON
EARTH.

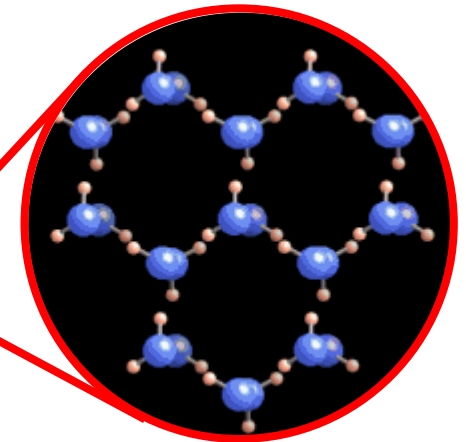
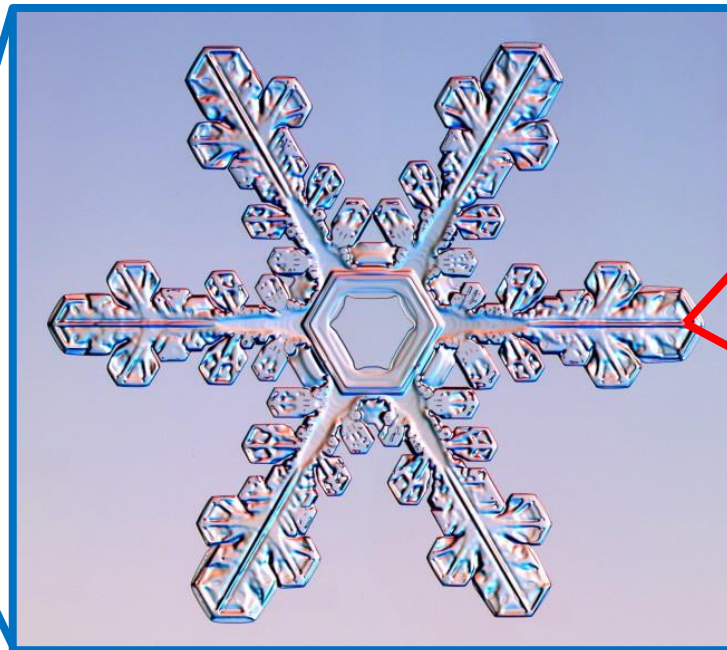
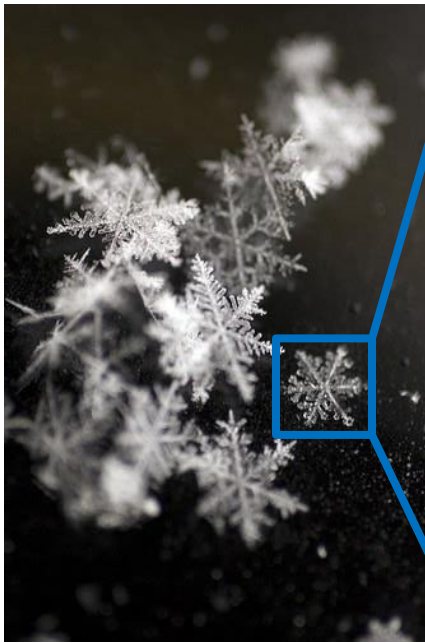
Atoms are very small!

Voyage into the World of Atoms:

https://www.youtube.com/watch?v=7WhRJV_bAiE

Snowflake ~1-3 mm

Ice crystal unit
cell 5 nm



Single atom
 $\sim 1\text{\AA} = 10^{-10}\text{ m}$

A typical **snowflake** is made of about 10^{18} - 10^{19} atoms.

Study of Matter

- **Physics** – *physical science* that studies forms of matter, its change and motion through space-time, and related concepts such as energy and force.
- **Chemistry** – *physical science* that studies material substances, their composition and change of composition (chemical reactions), as well as matter behavior related to chemical reactions.

Physical science
– branch of natural science that studies non-living systems.

Natural science – major branch of science, that tries to explain and predict nature's phenomena, based on empirical evidence.

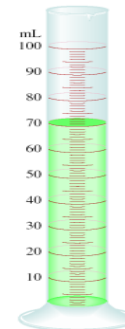
Science – systematic effort of acquiring knowledge—through observation and experimentation coupled with logic and reasoning.

Physical Properties of Matter

We can describe physical properties of matter in terms of physical quantities and laws.

- An extensive property **depends upon how much matter is being considered:**

- mass
- volume
- electrical charge



- An intensive property **does not depend upon how much matter is being considered:**

- density
- temperature
- color
- elasticity
- metallicity
- solubility etc...

