## Chemistry 0 Week 6 Assignment on 11/07/2021 Due Date: 11/13/2021

	PΙ	ease	answer	the	following	questions
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1.	A methane mo	olecule (CH₄)	is an example	of a covalent l	bond because
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- 2. Ionic bonds form between what types of elements?
- 3. Covalent bonds form between what types of elements?
- 4. Which of the below elements will most likely form an Ionic Bond?
  - a. P and O
  - b. S and O
  - c. Ca and Cl
  - d. Ne and CI
- 5. How many atoms are in CaCl<sub>2</sub>?
- 6. When Calcium forms a bond, what will occur?
- 7. Which of the below elements will most likely form a Covalent Bond?
  - a. Na and Cl
  - b. Al and Mg
  - c. H and O
  - d. K and Na
- 8. The force that holds two atoms together is called a \_\_\_\_\_.
- 9. How many electrons does chlorine need to gain in order to have a full outer shell?
- 10. A charged Particle is called an \_\_\_\_\_.