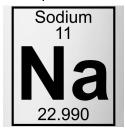
Chemistry 0 Week 2 Assignment on 10/03/2021

Please answer the following questions:

- 1. An oxygen atom contains eight protons, how many electrons does it have?
- 2. What is the atomic number of the element Sodium (Na)?
- 3. The element iron (Fe) has an atomic number of 26. How many protons does an atom of iron contain?

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- 4. The element scandium (Sc) contains 21 protons and 24 neutrons in its atom. What is its atomic number?
- 5. The element scandium (Sc) contains 21 protons and 24 neutrons in its atom. What is its mass number?
- 6. An atom has a mass number of 15, and it contains 8 protons, how many neutrons does it have?
- 7. Unknown element X has a mass number of 55 and contains 30 neutrons in its atoms. Identify element X.
- 8. The atomic weight of an element is the average weight of mixture of two or more _____?
- 9. In the picture below, what does 22.990 represent?



10. All atomic weights can be expressed in atomic mass unit (amu). 12 amu would equal the mass of a single ¹²C atom. One amu equal to what fraction of a single ¹²C atom?