

What is Energy?

Energy is defined as the ability to do work, that is, produce certain changes within a system.

Types (forms) of energy:

- Mechanical
 Chemical

 - Electromagnetic
- Heat (Thermal)
 Nuclear



- We cannot actually see energy ⁽²⁾
- We can observe how energy makes matter change in numerous ways (for example, change of physical properties, change of state, change of position etc.)
 - We can observe how energy changes its form.

Mechanical Energy

Energy due to an object's motion or position.









Chemical Energy

Chemical energy is an inherent energy of a substance due to its chemical composition:

All compounds are held together by chemical bonds.

 All types of chemical bonds have specific stored energy that can be released (transferred to another form, for example, heat or light) when the bonds are broken in a chemical reaction.





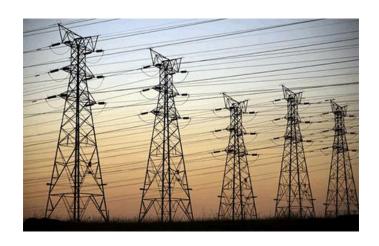






Electromagnetic Energy

Electric (from electric fields), Magnetic (from magnetic fields), Radiant (from electromagnetic radiation including light)











Thermal Energy

Thermal energy originates from the individually random, or disordered, motion of particles in a substance:

 All objects constantly give off or gain thermal energy.

 Heat is an amount of thermal energy being transferred in a given process in the direction of decreasing temperature.







Heat generated

from the

Nuclear Energy

Energy stored in the nucleus of an atom.

Low Speed Neutron

Neutron

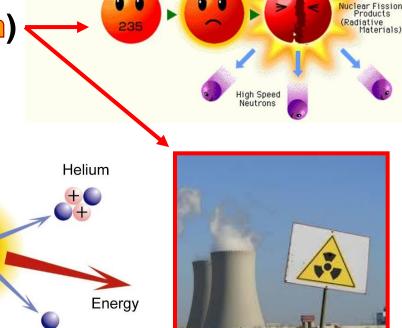
Nuclear energy is <u>released in</u> the form of heat and light when:

> the nucleus splits (fission)

> the nuclei collide at high speeds and join (fusion).

Deuterium

Tritium



(Fisson)

Nuclear energy is the most concentrated form of energy.