Length scales in Nature

1 mm

Grain of sugar, small insects, etc

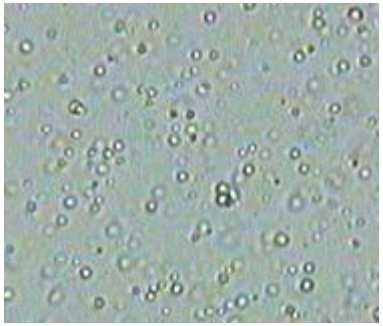
1 km

Brooklyn bridge

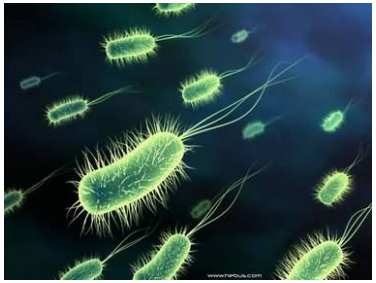
10^{-3} m 1 m 10^{3} m

1 micron (1μm)

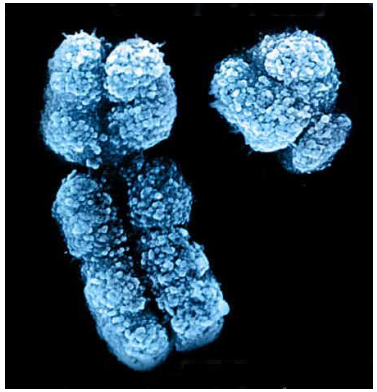
Particles in smoke, milk, etc
(1-20 μm)



Bacteria
(1-10 μm)



Human Chromosome
(2 -10 μm)



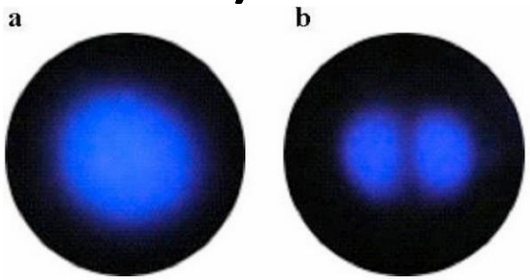
1000 km

10^{-6} 10^{-3} 1 m 10^{3} 10^{6}

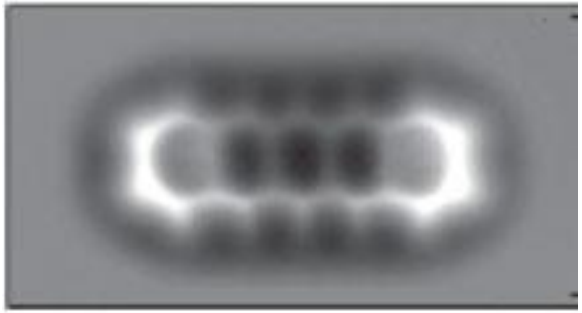
1 nanometer = 10 Angstrom
(1 nm =10 Å)

1,000,000 km
(3 light seconds)

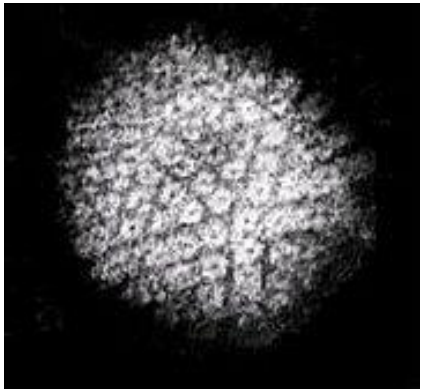
Atom (1 Å)

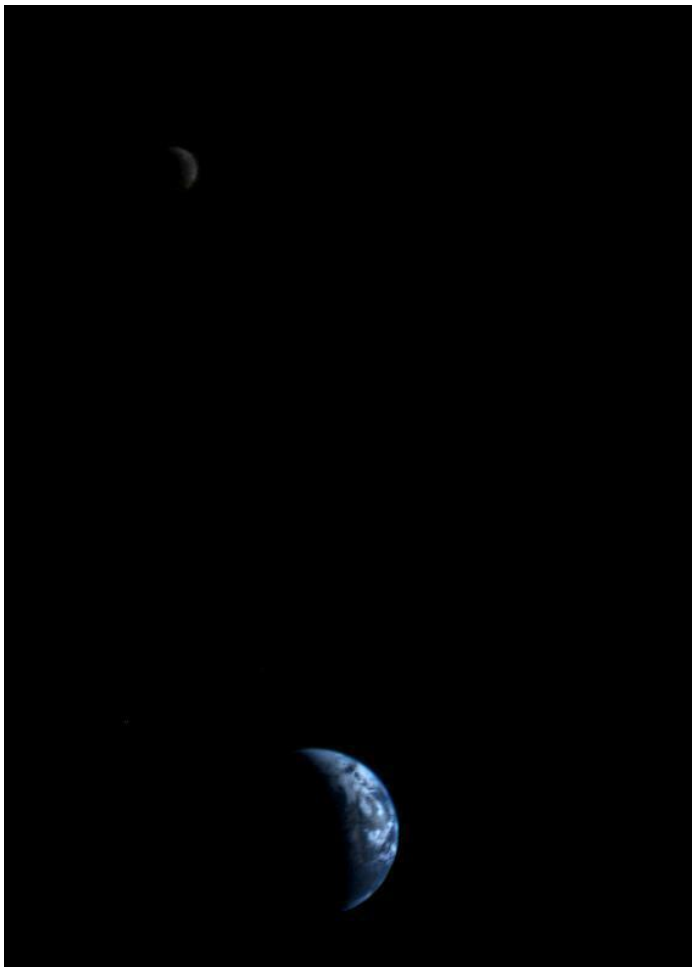


Molecule
(1nm)



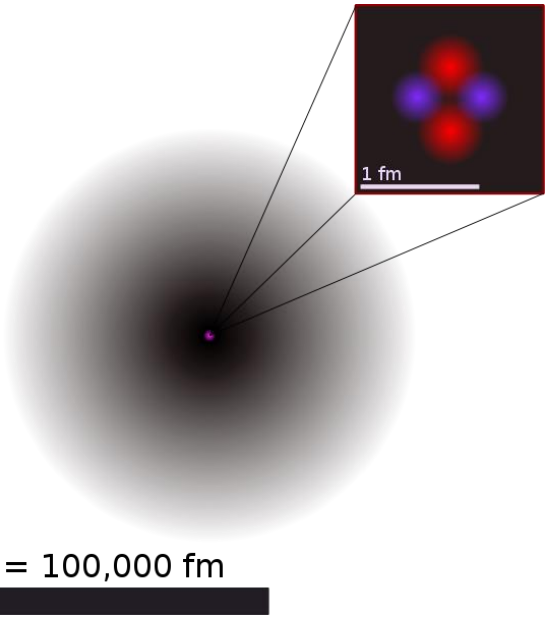
Virus (>10 nm)





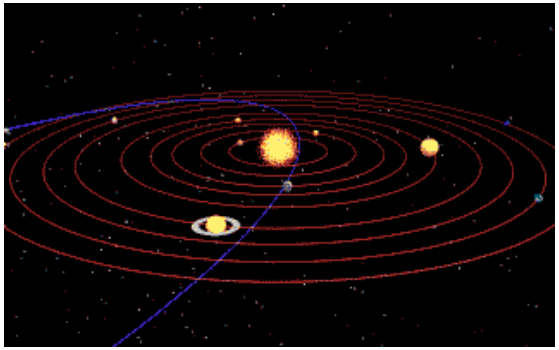
10^{-9} 10^{-6} 10^{-3} 1 m 10^{3} 10^{6} 10^{9}

Proton, neutron, atomic nucleus

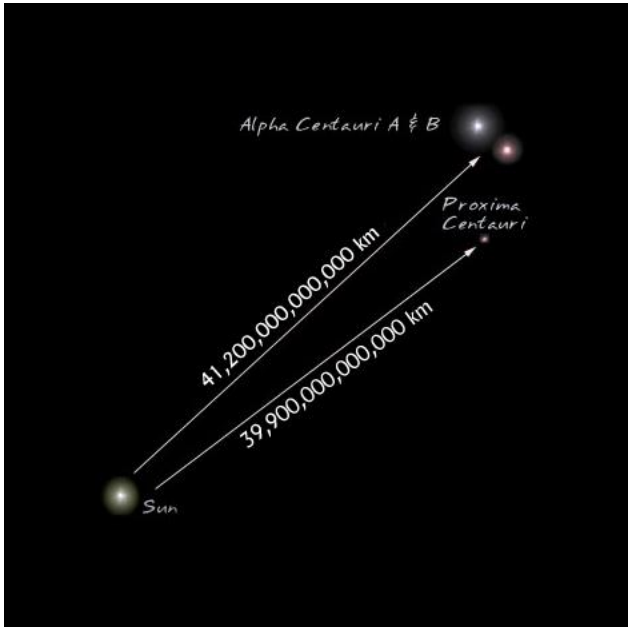


1 Å = 100,000 fm

10^{12} m = 1 billion km ≈ 1 light hour



10^{16} m ≈ 1 light year



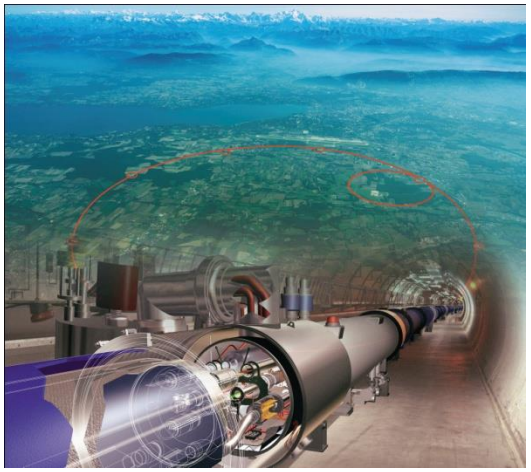
10^{-15} 10^{-12} 10^{-9} 10^{-6} 10^{-3} 1 m 10^{3} 10^{6} 10^{9} 10^{12} 10^{15}

Modern Physics

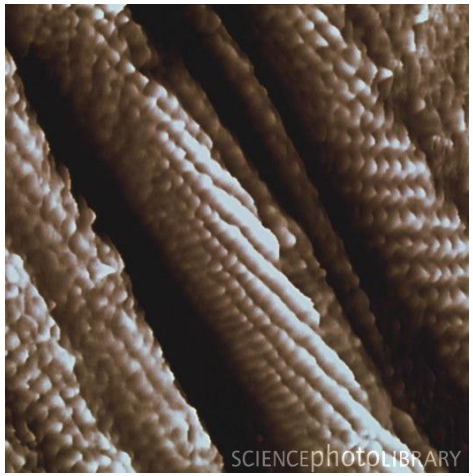
High Energy Physics

Condensed Matter Physics

Astrophysics & Cosmology



Large Hadron Collider (LHC)



Carbon Nanotubes

Our Galaxy (Milky Way)

?

10^{-35} 10^{-21} 10^{-15} 1 m 10^{15} 10^{21} 10^{27}

Homework 1

Problem 0.
Watch the classic documentary called “Powers of Ten”
https://www.youtube.com/watch?v=0fKBhvDjuy0 (you can also easily google it)
Please go through length scales of various objects.
In addition to the classroom presentation, you might want to use this website:
http://micro.magnet.fsu.edu/primer/java/scienceopticsu/powersof10/

Problem 1.

Estimate the number of atom in a grain of salt. Assume the grain to be a cube 1x1x1 mm, and each atom to be a cubic brick.

Problem 2.

Estimate the number of cells in your body, if a typical human cell is about 10 micron in size.