

# Length scales in Nature

1 mm



Grain of sugar, small insects, etc

1 km



Brooklyn bridge

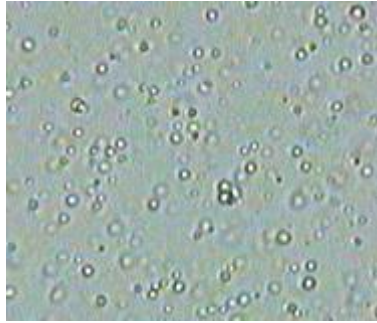
$10^{-3}$  m

1 m

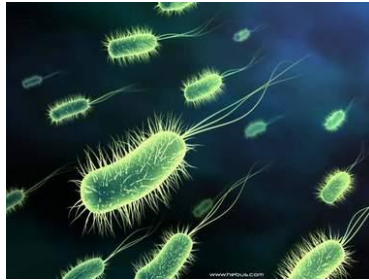
$10^3$  m

1 micron (1 $\mu$ m)

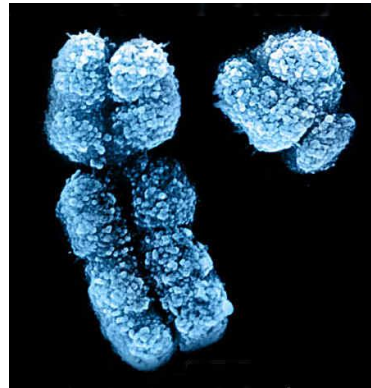
Particles in smoke, milk, etc  
(1-20  $\mu$ m)



Bacteria  
(1-10  $\mu$ m)



Human Chromosome  
(2 - 10  $\mu$ m)



1000 km



10<sup>-6</sup>

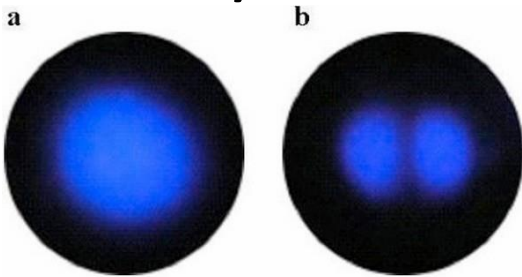
10<sup>-3</sup>

1 m

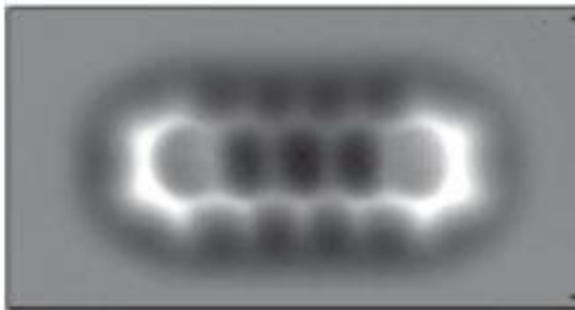
10<sup>3</sup>

10<sup>6</sup>

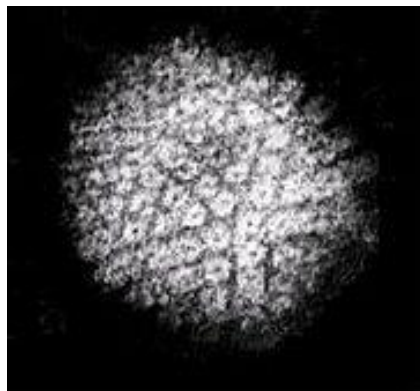
1 nanometer = 10 Angstrom  
(1 nm = 10 Å)



Atom (1 Å)



Molecule  
(1nm)



Virus (>10 nm)

1,000,000 km  
(3 light seconds)



$10^{-9}$

$10^{-6}$

$10^{-3}$

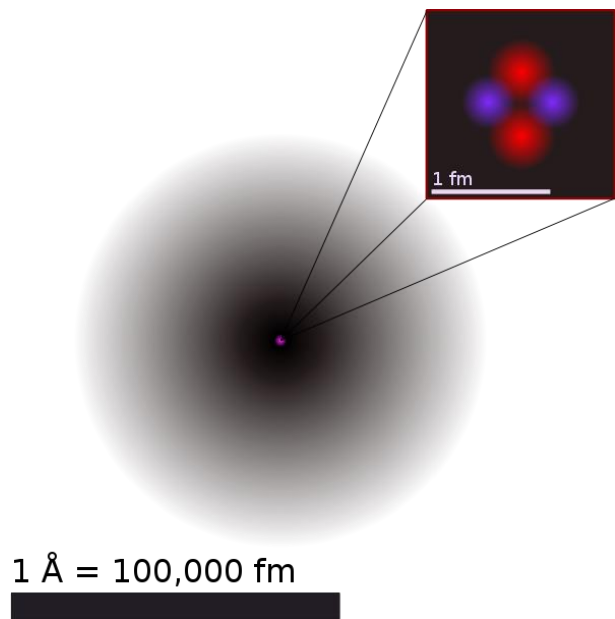
1 m

$10^3$

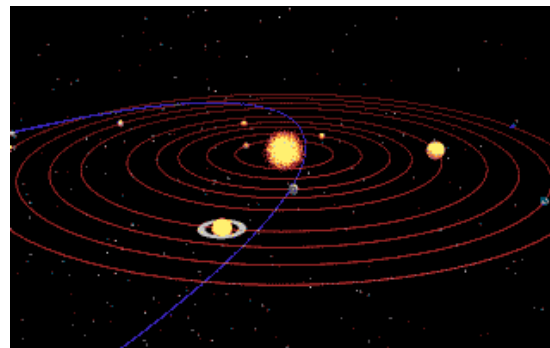
$10^6$

$10^9$

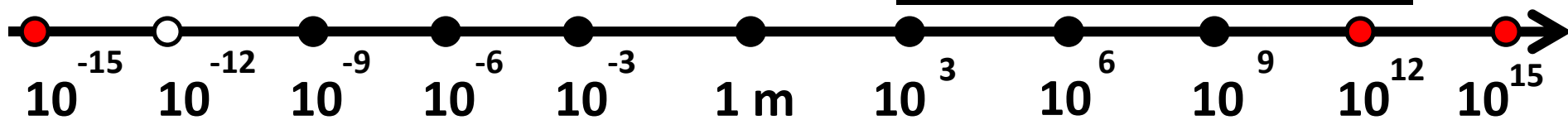
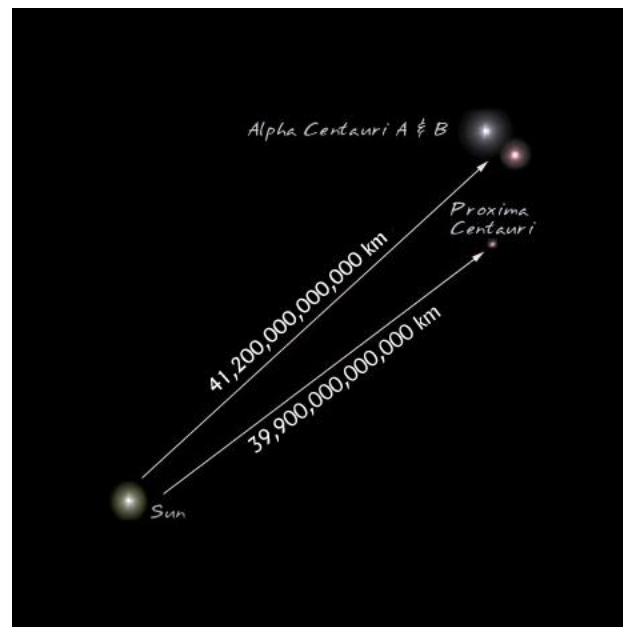
# Proton, neutron, atomic nucleus



$10^{12}$  m = 1 billion km  $\approx$  1 light hour



$10^{16}$  m  $\approx$  1 light year

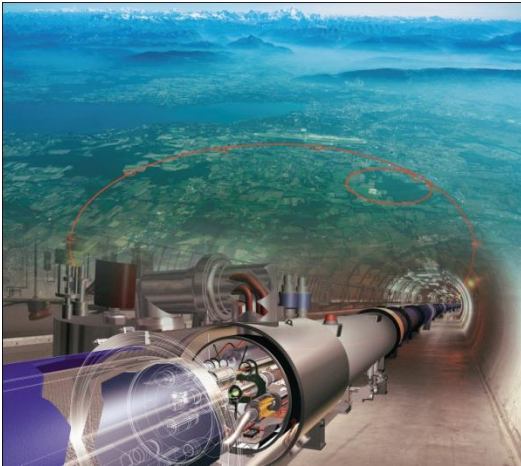


# Modern Physics

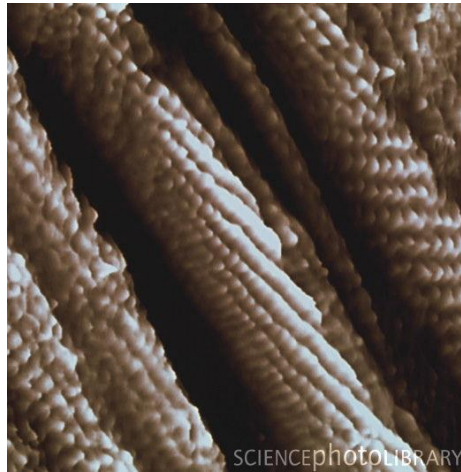
High Energy Physics

Condensed Matter Physics

Astrophysics & Cosmology



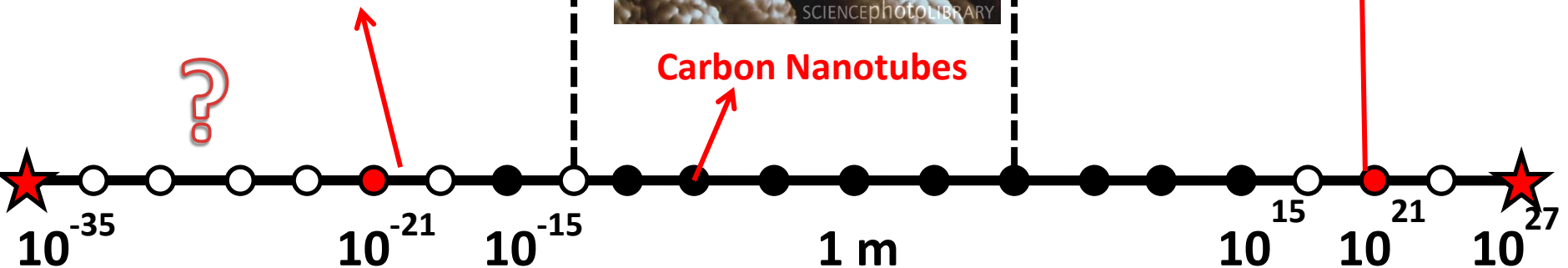
Large Hadron Collider (LHC)



Carbon Nanotubes



Our Galaxy (Milky Way)



# Homework

Water molecule can be approximated as a sphere of radius  $2 \text{ \AA}$  ( $1 \text{ \AA} = 10^{-10} \text{ m}$ , is called Angstrom). Estimate how many molecules is there in one glass ( $200 \text{ cm}^3$ ) of water.