| 2.1 Notes Name | | Date: |
|---|---|-------------------------------------|
| Warm Up: Silently Draw a Bohr Diagra | am for each of the following elem | ents: |
| <u>Lithium</u> | <u>Sodium</u> | <u>Potassium</u> |
| | | |
| Electron Configuration: | Electron Configuration: | Electron Configuration: |
| The Periodic Table of Element | ts | |
| As you watch the video, record notes | below and try to fill out the key | points that are listed. |
| | | |
| | | |
| | | |
| Key Understandings * Gold, silver, and copper are easily sp * Alchemist Graham isolated the elem * Lithium, Sodium, and Potassium all * The current model of the Periodic Table | ent Phosphorus from his react similarly because they are i | n the same |
| Vocabulary | | |
| Element | | |
| | | |
| The Periodic Table of Elements is a veelements found on Earth. | ery strategic arrangement of all of | f the known, an even some unknown |
| As you watch the next video, pay very how important Mendeleev's contribut | | ry used and try and understand just |
| Periodic Table Vocab | | |
| <u>Periodic</u> | | 1 |
| <u>Period</u> | | |
| Current | | |
| Group | | |
| Flemental Families | | |

| Look back at your warm up. What do all of these elements have in common? | | |
|---|--|--|
| Look at the Periodic Table projected on the board. Anything interesting about these three elements? | | |
| Group 1 Metals - also known as the Metals all have | | |
| | | |
| | | |
| When elements are in the same group on the periodic table | | |
| Video: As you watch the video, record your observations paying particular attention to what is similar and different about the Group 1 metals when they react with water. | | |
| Exit Ticket: | | |