The chemical formula of water is $\mathrm{H}_{2} \mathrm{O}$. This means that the water molecule is made up of
a) Two hydrogen atoms and one oxygen atom
b) Two oxygen atoms and one hydrogen atom
c) Two hydrogen atoms and two oxygen atoms
d) Two molecules of water

Water can dissolve sodium chloride because
a) The polar areas of water molecules attract the oppositely charged ions of sodium chloride
b) The shape of the water molecules pushes the sodium and chloride ions apart
c) The oxygen in the water molecules reacts with the sodium and chloride ions
d) Water molecules and sodium chloride are covalently bonded

## True or False?

Water is a polar molecule.

Calculate 1 mole of $=$ $\qquad$
$\mathrm{C}_{2} \mathrm{H}_{6}$

NaOH
$\mathrm{CuSO}_{4}$
$\mathrm{C}_{3} \mathrm{H}_{8} \mathrm{O}$

