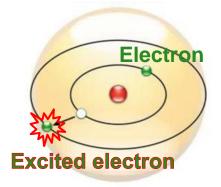
Gases

VS

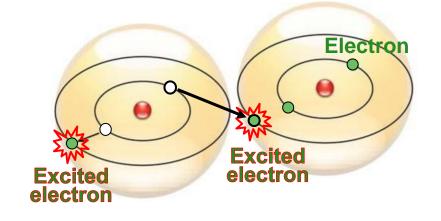


Solids/Liquids

atoms close to each other



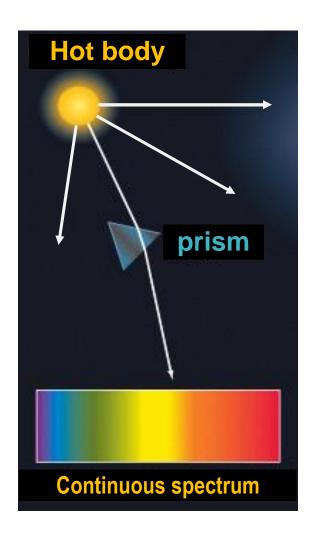
much greater number of possible colors



Thermal Radiation

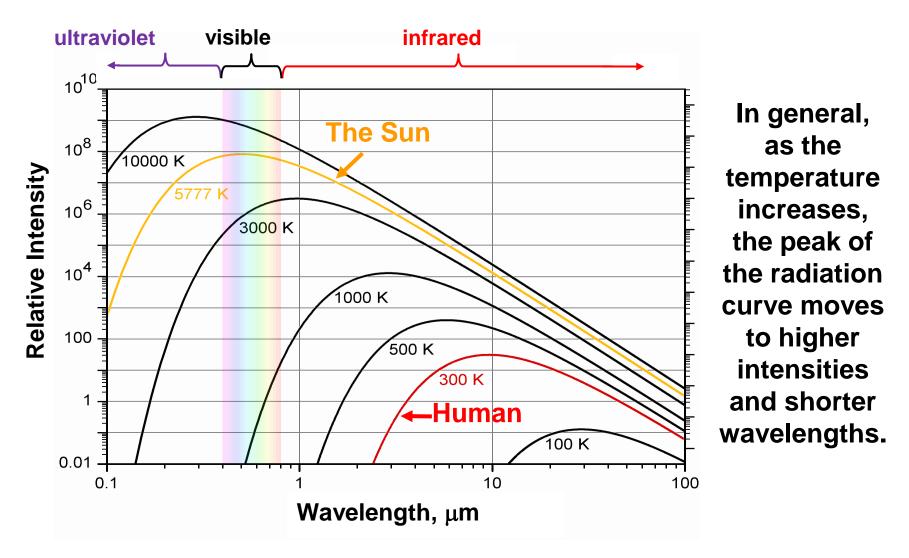
All normal matter emits electromagnetic radiation when it has a temperature above absolute zero.

- This radiation represents a conversion of a body's thermal (heat) energy into electromagnetic energy, and is therefore called thermal radiation.
- When the atoms are in a <u>condensed state</u> (solid or liquid matter), the "hot" electrons can make transitions not only within the energy levels of their own atom, but also <u>between the levels of neighboring atoms</u> (that can be of same or different kind).
- This results in a much larger number of possible transitions with corresponding frequencies of radiant energy, producing a continuous color spectrum.



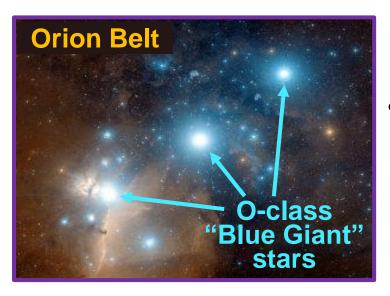
Thermal Radiation Spectrum

The <u>exact thermal radiation spectrum</u> depends upon properties of the material and the temperature.



Everything Glows!

 The temperature at which all solids glow a dim red is about 800 K (over 500°C or 900°F).



 People are emitters of light in the infrared region (peak ~9.5μm).

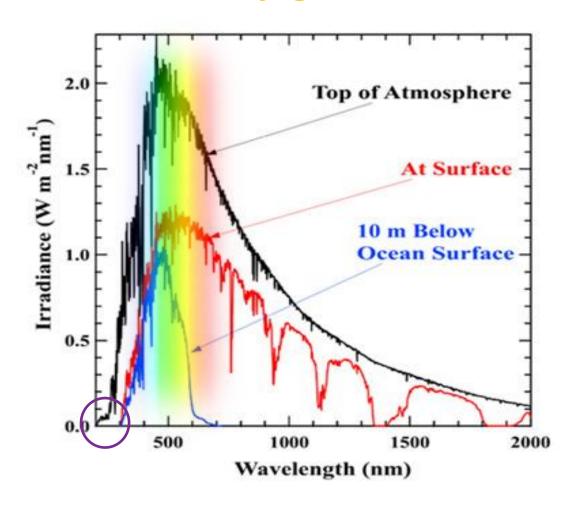


A very hot object (10,000 K)
would emit a significant amount
of energy in the ultraviolet and
x-ray region of the spectrum.



Sunlight

The Sun emits EM radiation across most of the electromagnetic spectrum. On Earth, solar radiation is obvious as daylight when the Sun is above the horizon.

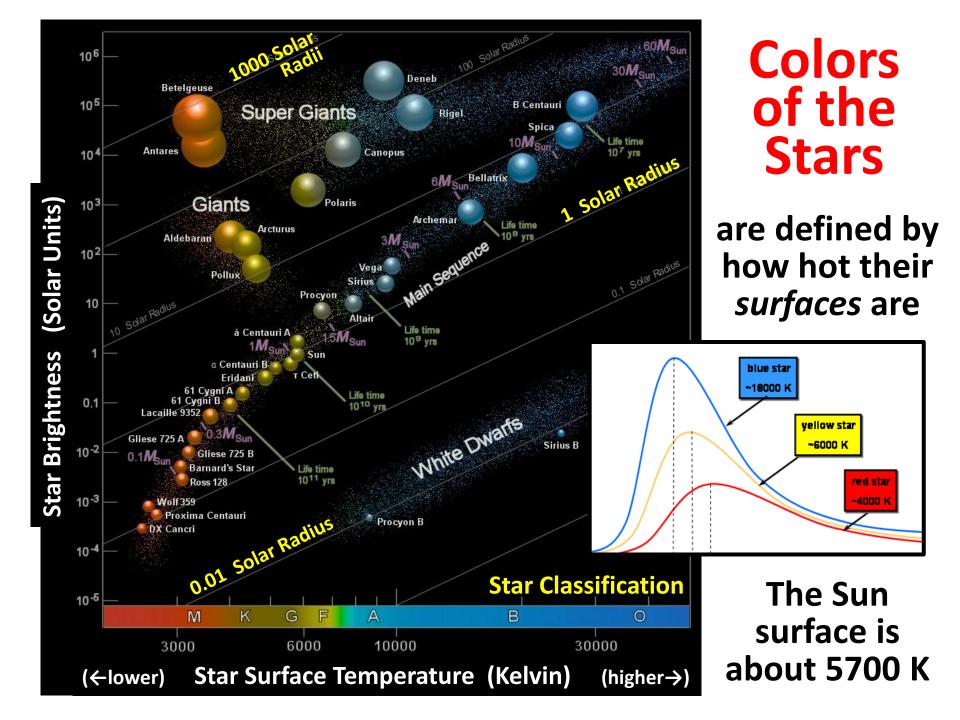


Sunlight composition at top of atmosphere:

- ~9% ultraviolet radiation
- > ~40% visible light
- > ~51% infrared radiation
- a <u>very small amount of</u> <u>extreme ultraviolet and</u> <u>X-ray radiation</u> from solar corona

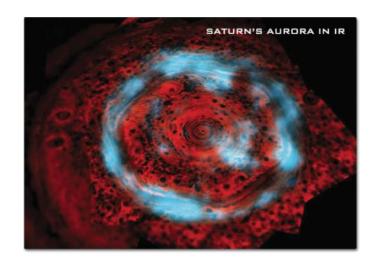
Filtered Sunlight at Earth's surface:

- ~5% ultraviolet radiation
- > ~43% visible light
- ~52% infrared radiation



"Seeing" the Invisible with Infrared





From elusive leopards...

visible infrared

...to hiding young stars!