Problems:

1. Each of two cylinders contain same mass of gas. Pressure in one cylinder is $4,000 \mathrm{~Pa}$, pressure in the other $6,000 \mathrm{~Pa}$. Find the pressure which will established if we connect the cylinders with a narrow tube (neglect the volume of the tube).
2. What is the density of nitrogen at $\mathrm{T}=0^{\circ} \mathrm{C}$ and pressure of $10^{5} \mathrm{~Pa}$.
3. After we let a certain amount of gas out of the cylinder, the pressure in the cylinder dropped by $40 \%$ and the temperature - by $20 \%$. Find the fraction of the initial gas mass which was lost. 3
