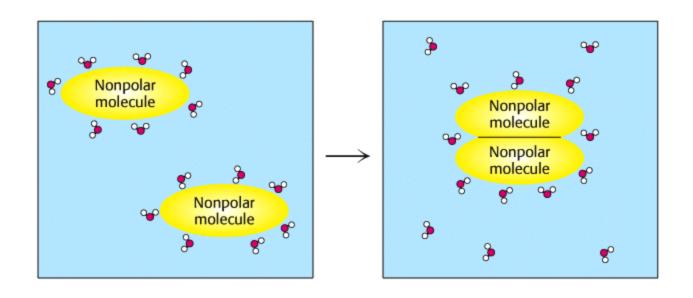
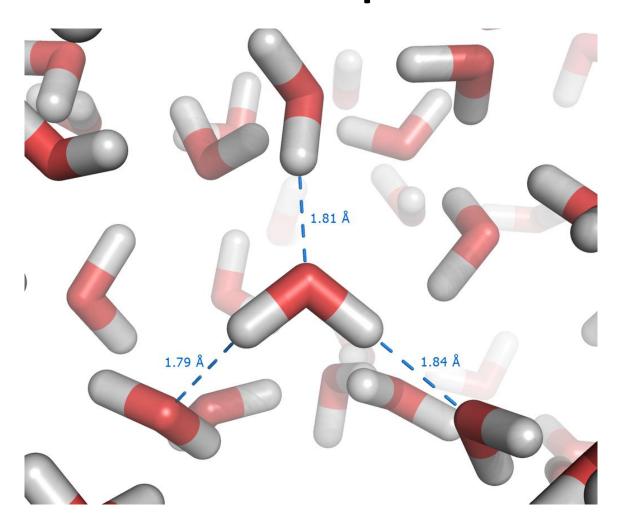
Hydrophobic interactions and biomembranes.

Hydrophobic effect

Non-polar molecules aggregate in aqueous solutions in order to separate from water.



Dynamic hydrogen bonds between molecules of liquid water



Hydrophobic Effect

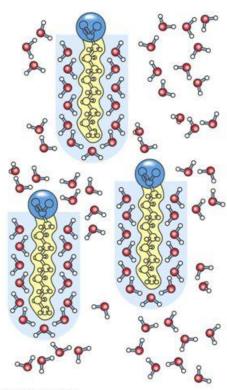
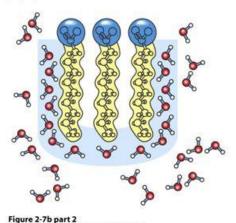


Figure 2-7b part 1
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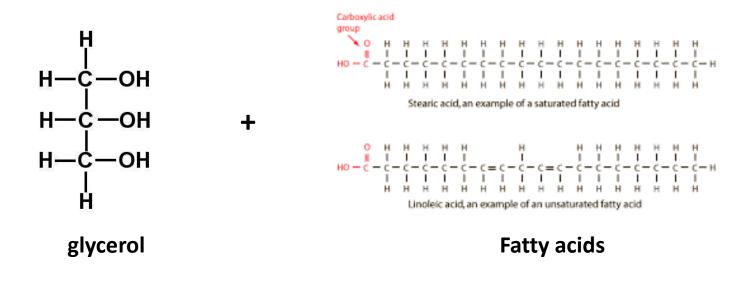
- Lipid molecules disperse in the solution; nonpolar tail of each lipid molecule is surrounded by ordered water molecules
- Lipid aggregates Water released, surface area reduced

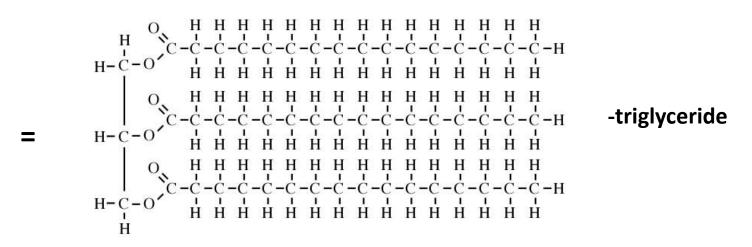




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Example of a hydrophobic molecule – triglyceride (fat)



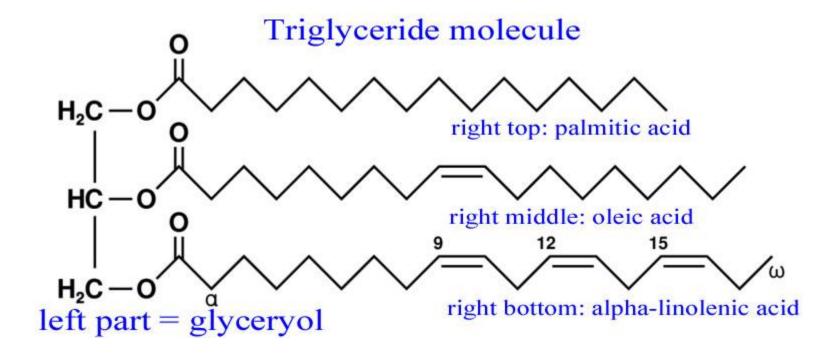


Ester bond

 Ester chemical bond is a covalent bond derived from substitution reaction of a carboxylic acid and an alcohol.

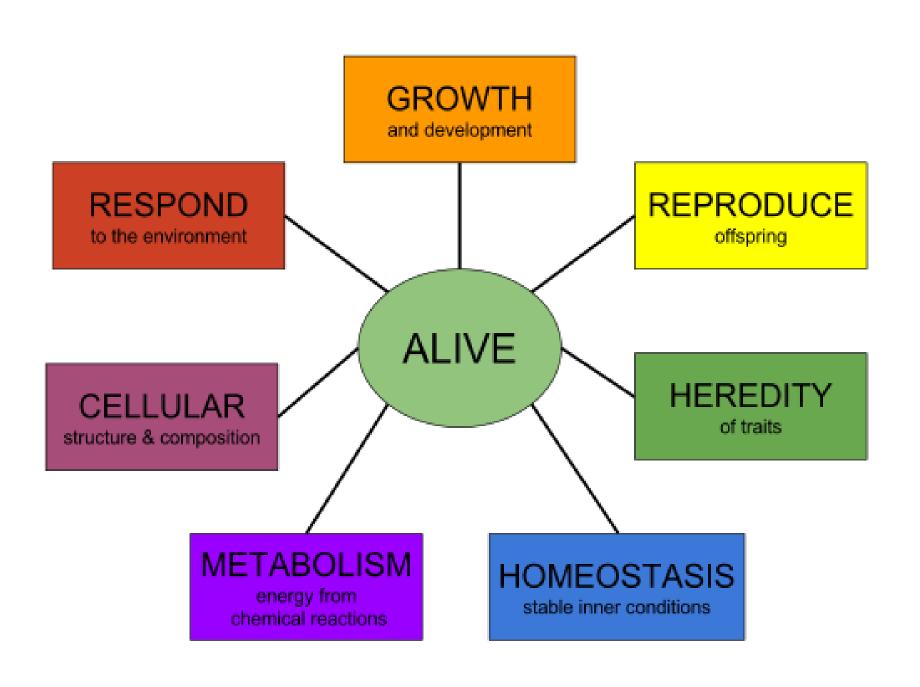
$$R_1-C$$
 + $HO-R_2$ $\xrightarrow{H+/heat}$ R_1-C + H_2O O- R_2 Ester Water

triglyceride

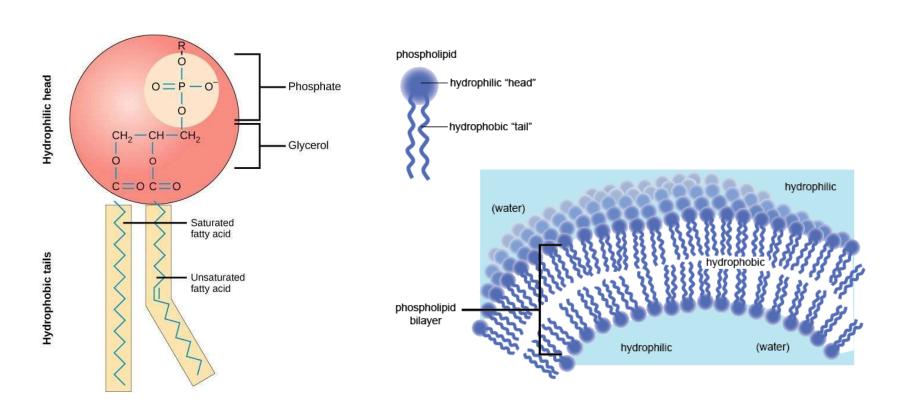


Cell membrane

- A cell is surrounded by the cell membrane that separates its interior from the outside environment (the extracellular space).
- The cell membrane is the barrier that keeps ions, proteins and other molecules where they are needed and prevents them from diffusing into areas where they should not be.
- Eukariotic cells have internal compartments separated from the rest of the cell by their own membranes



Cell membrane consists of lipid bilayer



• The cell membrane is selectively permeable and able to regulate what enters and exits the cell, thus facilitating the transport of materials needed for survival. The movement of substances across the membrane can be either "passive", occurring without the input of cellular energy, or "active", requiring the cell to expend energy in transporting it. The cell membrane thus works as a selective filter that allows only certain things to come inside or go outside the cell.

A detailed diagram of the cell membrane

