$$
\text { Density }=\frac{\text { Mass }}{\text { Volume }} \quad \text { or } \quad \rho=\frac{M}{V}
$$

## Homework Problem.

The table below shows masses ( $\mathbf{m}$ ) and dimensions (radius $\mathbf{R}$, thickness $\mathbf{d}$ ) of various US coins.
a) Derive a general formula for density $\rho$ of the material from which a coin is made of. It should look like $\rho=$ "some mathematical expression of $m, R$, and $d$ ".
b) Based on your formula, fill the missing densities in the table (in $\mathrm{g} / \mathrm{cm}^{3}$ ). Pay attention to units! Based on your results, could a penny be made of Copper? Nickel of Nickel?

| Coin | penny | nickel | dime | quarter |
| :---: | :---: | :---: | :---: | :---: |
| $R(\mathrm{~cm})$ | 0.95 | 1.06 | 0.90 | 1.53 |
| $\mathrm{~d}(\mathrm{~mm})$ | 1.55 | 1.95 | 1.35 | 1.75 |
| $\mathrm{~m}(\mathrm{~g})$ | 2.5 | 5.0 | 2.27 | 5.67 |
| $\rho\left(\mathrm{~g} / \mathrm{cm}^{3}\right)$ |  |  |  |  |

